

Is it Possible to Regain Lost Minerals to Initial Enamel Lesions Using Enamel Matrix Derivatives Combined with Casein Phosphopeptide Amorphous Calcium Phosphate: An *in Vitro* Study

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ABSTRACT

Objective: This study aimed to assess the efficacy of enamel matrix derivatives (EMD) used in combination with casein phosphopeptide amorphous calcium phosphate (CPP-ACP) to recover minerals to the enamel after artificial caries formation.

Methods: Fifty enamel samples were prepared from the buccal surfaces of extracted human third molars. The samples were divided into five groups: (G1) Untreated enamel samples (control), (G2) Application of 500 ppm sodium fluoride (NaF), (G3) CPP-ACP, (G4) EMD, and (G5) CPP-ACP + EMD. All of the samples were placed in an acidic buffer solution for 96 hours to simulate a carious lesion. Ca/P ratios were calculated using energy dispersive X-ray spectroscopy (EDX). After seven days of the remineralization procedure, the mineral contents of the samples were re-measured. All the data were analyzed statistically.

Results: There were no statistically significant differences in the mineral contents of the samples between the groups after demineralization ($p > 0.05$). The Ca/P ratios of G2, G3, G4, and G5 increased significantly ($p < 0.05$) after remineralization. The highest levels of the Ca/P ratio were obtained in G5.

Conclusion: Despite the limitations of this *in vitro* study, the combined use of CPP-ACP and EMD may increase the remineralization potential. Furthermore, this procedures may be an alternative for providing enamel remineralization in future clinical trials.

Keywords: Casein phosphopeptide amorphous calcium phosphate, enamel matrix derivatives, enamel remineralization, fluoride, SEM-EDX

INTRODUCTION

Dental caries is an infectious disease of the teeth that results in the dissolution and destruction of hard tissues such as the enamel and dentin (1). The first sign of dental caries is the initial caries lesion defined as “subsurface enamel porosity from carious demineralization” (2), and it can be reversed by the recovery of minerals to the lost structure (3). Some methods or materials like the application of topical fluoride or CPP-ACP provided aided remineralization (4). Fluoride is a classic anti-caries agent, although it is more effective for sound enamel than caries lesions (5) since its effectiveness is limited by the availability of calcium and phosphate ions (6). CPP, which is the protective factor in milk, was obtained by the digestion of casein with trypsin enzyme, using the selective precipitation method (7). CPP can stabilize calcium phosphate as a CPP-ACP complex (8), and these complexes promote remineralization by increasing calcium phosphate in the dental plaque (9).

More recently, biomimetic remineralization approaches are receiving increasing attention as a promising anti-caries therapy (10). The biomimetic synthesis of materials such as enamel can provide a noninvasive alternative treatment for early carious lesions. Amelogenin is the most common enamel matrix protein that plays a role in the formation and growth of the enamel crystal structure (11). Regarding the interaction of amelogenin and calcium phosphate, it is believed that amelogenin modulates the calcium phosphate nanocrystalline structure and plays an important role in enamel biomineralization (11). Amelogenin, calcium, and phosphate ions are important substances for the formation of organized hydroxyapatite crystals *in vitro* (12). Enamel matrix derivatives (EMD) are commercially available derivatives of enamel matrix proteins (EMP). EMD includes 90% amelogenin and 10% pig enamel matrix protein derivatives, which have been shown to promote periodontal ligament cell proliferation and collagen production, and also to enhance *in vitro* mineralization (13).

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In light of these knowledge, the null hypothesis of the present *in vitro* study was that CPP-ACP and EMD would have a positive synergistic effect on the recovery of the lost structure of the enamel surface. This *in vitro* study aimed to evaluate the remineralization potential of CPP-ACP and EMD in initial caries lesions by EDX and make comparisons between the groups.

METHODS

Preparation of the Enamel Samples

Fifty impacted human third molar teeth were collected for this study, and ethics committee approval was obtained from the İnönü University/Turkey Clinical Research Ethics Committee (2013/146). The teeth were stored in 4°C deionized water containing 0.2% thymol for at most 30 days. The samples were evaluated using a stereomicroscope (Nikon, Tokyo, Japan) at 2× magnification to choose the sound enamel surfaces. The teeth were then cut using the IsoMet™ Low Speed Precision Cutter (Buehler, Germany). The enamel surface of the blocks was polished on a polishing machine (Ecomet 3, Bueller, IL, USA) and water-cooled to obtain a smooth enamel surface. Then, they were covered with two coats of acid-resistant nail varnish, except for a 3×3 mm² enamel window.

Caries Lesion Formation

Early artificial caries lesions were produced in 50 enamel samples. The samples were placed in a demineralization solution containing 2.2 mM CaCl₂, 2.2 mM NaH₂PO₄, 50 mM acetic acid, and 1 M KOH. The pH of the solution was adjusted to 5.0, and the demineralization was performed at 37°C for 96 hours (14).

pH-Cycling Model

The samples were divided into five groups: (G1) Untreated enamel samples (control), (G2) 500 ppm sodium fluoride (NaF, Oral-B Stages, Oral-B Laboratories, Netherlands), (G3) CPP-ACP (Tooth Mousse, containing 10% CPP-ACP; GC Int., Tokyo, Japan), (G4) EMD gel (Emdogain, lyophilized protein fractions dissolved in acetic acid, Straumann, Biora, Sweden), and (G5) CPP-ACP + EMD. pH-cycling was carried out using a remineralization solution (1.5 mM CaCl₂, 0.9 mM NaH₂PO₄, 150 mM KCl, 1 M KOH, and pH 7.0) and demineralization solution (2.2 mM CaCl₂, 2.2 mM NaH₂PO₄, 50 mM acetic acid, and 1 M KOH, pH 5.0) (15). The experimental design included 2 demineralization/remineralization cycles a day for 10 days: immersion of the enamel samples in 10 mL of demineralizing solution, twice daily (at 9 AM and 9 PM) for 30 minutes; application of the solutions (described above) for 4 minutes (in group 5; CPP-ACP was applied for 4 minutes firstly, washing in distilled water for 10 seconds, then EMD gel was applied for 4 minutes); washing in

distilled water for 10 seconds; immersion in 100 mL of the remineralizing solution (16).

SEM-EDX Analysis

SEM-EDX analysis was performed to determine the mineral content of the enamel samples after caries lesion formation and pH-cycling. Elemental analysis was performed on both halves of the exposed enamel surfaces in 20 Kv and fluoride (F) %, calcium (Ca) %, and phosphate (P) % values were recorded atomically. The Ca/P% ratio was then calculated. Image analysis was also performed using an SEM to determine the structure of the initial carious lesions and the variations after remineralization.

Statistical Analysis

After demineralization and application of the remineralization agents to the demineralized samples, a paired t-test was performed for each group to evaluate the mineral changes in the samples. The one-way ANOVA test was conducted to compare mineralization differences between the groups after demineralization and remineralization. In addition, the post-hoc Tukey test was used to compare the differences between the groups after remineralization.

RESULTS

The mean atomic values of the F%, Ca%, and P% values were recorded atomically using SEM-EDX, and the Ca/P% ratios of all the study groups were statistically compared (Table 1).

The SEM-EDX analysis method was used after applying the demineralization procedure, and it revealed that there was no statistically significant difference between the groups in terms of the mean atomic Ca/P% ($p < 0.05$). After the remineralization, the mean atomic Ca/P% ratio in all groups, except for the control group, exhibited a statistically significant increase ($p < 0.05$). According to the paired t-test, the highest Ca/P% ratio after treatment was obtained in group 5, whereas the lowest Ca/P% ratio was obtained in group 1 (Table 2).

Due to the intergroup comparison of the mean Ca/P% ratios after remineralization using the post-hoc Tukey test, the mean Ca/P% ratio of group 5 was found to be higher than that of group 2. However, the difference between these values was not found to be statistically significant ($p > 0.05$). The mean Ca/P% ratio obtained from groups 2 and 5 was higher than the mean Ca/P% ratio obtained from group 3, and the mean Ca/P% ratio obtained from group 3 was higher than that obtained from groups 1 and 4 ($p < 0.05$). The mean Ca/P% ratio obtained from group 4 was higher than that obtained from group 1. However, the difference between these values was not found to be statistically significant ($p > 0.05$; Table 3).

By evaluating the SEM images according to the treatment groups, we observed the enamel surfaces of the groups (Figure 1). The small particles and cracks found in the control groups (a) could be due to the dissolved Ca/P ions from the enamel subsurface. Prism structure could be seen in NaF (b) and CPP-ACP+EMD (e) with smaller pores and showed a dense packed precipitation layer. CaF₂ deposition was also observed

Main Points:

- The combined use of CPP-ACP and EMD increased Ca/P% ratio at the enamel surface.
- CPP-ACP showed close efficacy to the fluoride in enamel remineralization.
- The use of EMD alone was not sufficient to reverse the mineral loss from the enamel.

Table 1. Comparison of the mean atomic values of the F%, Ca%, P%, and Ca/P% ratios in the demineralization and remineralization periods

Elements %		Groups					OWA P
		G1	G2	G3	G4	G5	
F	Demineralization	13,44±1,29	13.44±0.84	13.57±1.05	13.45±1.06	13.59±0.84	0.054 0.99
	Remineralization	13,61±1,32	17.10±0.48	12.60±1.19	15.25±0.78	15.21±1.10	37.862*
Ca	Demineralization	15,53±1,79	15.03±1.73	14.37±0.94	14.23±0.89	14.06±1.46	1.887 0.12
	Remineralization	15,43±0,83	17.81±0.29	20.81±1.37	15.59±1.11	18.29±1.26	44.22*
P	Demineralization	8,74±0,81	8.56±0.85	8.34±0.30	8.38±0.73	8.09±0.56	1.305 0.28
	Remineralization	8,63±0,61	8.15±0.23	10.03±0.94	8.40±0.72	8.29±0.77	11.923*
Ca/P	Demineralization	1,77±0,10	1.75±0.09	1.72±0.08	1.70±0.10	1.73±0.09	0.90 0.46
	Remineralization	1,79±0,08	2.18±0.06	2.07±0.12	1.86±0.09	2.21±0.11	37.86*

Group 1: Control, Group 2: NaF, Group 3: CPP-ACP, Group 4: EMD, Group 5: CPP-ACP+EMD.

OWA: One-way Anova,

*: statistically significant difference (p<0.05)

Table 2. Comparison of the mean atomic values of the Ca/P% ratios in the demineralization and remineralization periods

	GROUPS					OWA P
	G1 n=10	G2 n=10	G3 n=10	G4 n=10	G5 n=10	
Demineralization	1.77±0.10	1.75±0.09	1.72±0.08	1.70±0.10	1.73±0.09	0.90 0.46
Remineralization	1.79±0.08	2.18±0.06	2.07±0.12	1,86±0,09	2.21±0.11	37.86*
Difference	+0.01±0.14	+0.42±0.12	+0.35±0.16	+0,15±0,14	+0.47±0.14	
PTT (P)	0.801	0.0001*	0.0001*	0.007*	0.0001*	

Group 1: Control, Group 2: NaF, Group 3: CPP-ACP, Group 4: EMD, Group 5: CPP-ACP+EMD.

OWA: One-way Anova, PTT: Paired T-Test,

*: statistically significant difference (p<0.05)

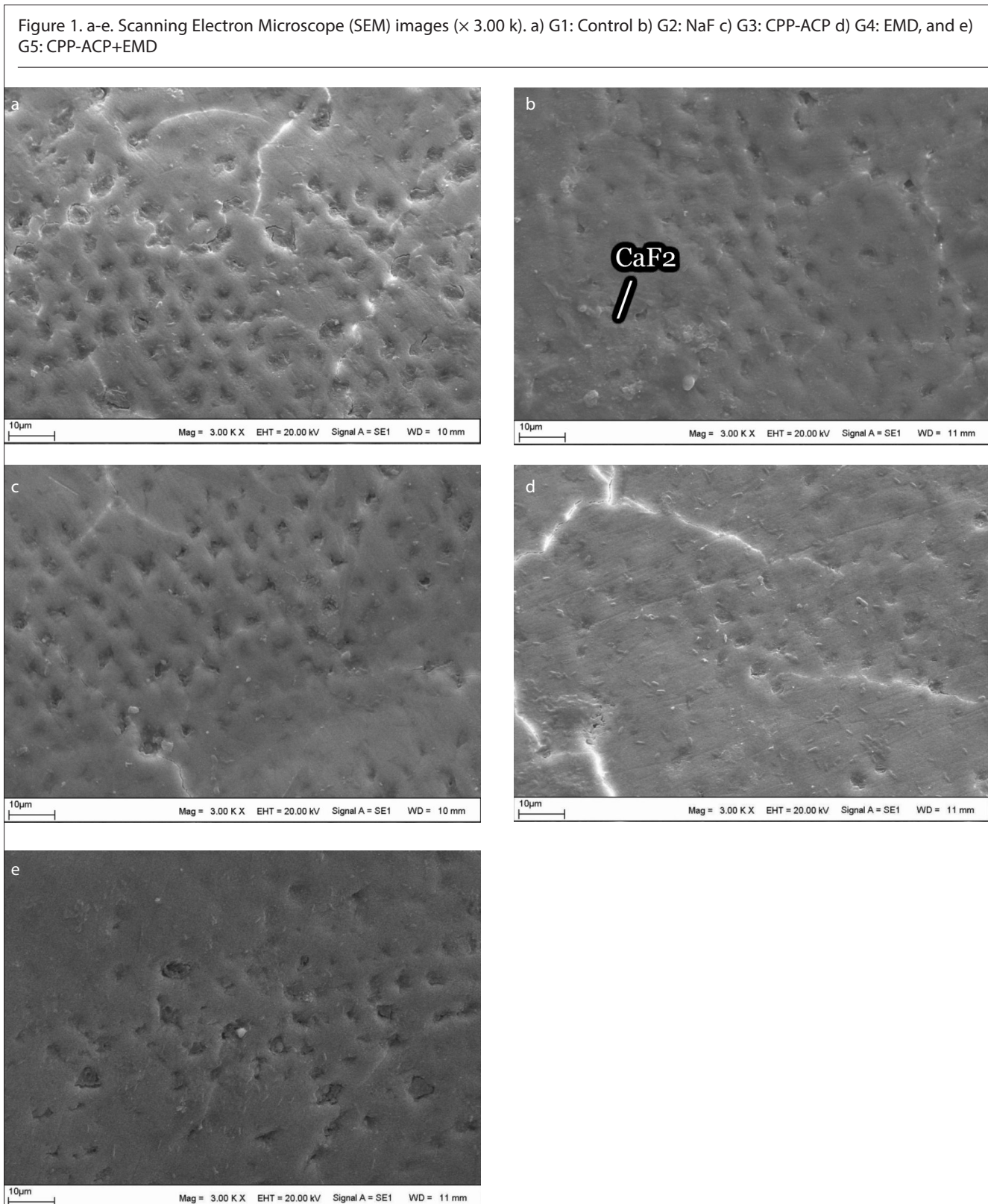
Table 3. Comparison of the mean atomic values of the Ca/P% ratios between the groups after remineralization

Group	Group	PHT(P)	Group	Group	PHT(P)	Group	Group	PHT(P)
G1	G3	0.0001*	G2	G3	0.014*	G3	G4	0.0001*
	G4	0.118		G4	0.0001*		G5	0.003*
	G5	0.0001*		G5	0.556		G2	0.014*
	G2	0.0001*		G1	0,0001*		G1	0.0001*
G4	G3	0.0001*	G5	G3	0.003*			
	G5	0.0001*		G4	0.0001*			
	G2	0.0001*		G2	0.556			
	G1	0.118		G1	0.0001*			

Group 1: Control, Group 2: NaF, Group 3: CPP-ACP, Group 4: EMD, Group 5: CPP-ACP+EMD

PHT: Post Hoc Tukey,

*: statistically significant difference (p<0.05)



in the NaF (b) group, but CPP-ACP+EMD (e) seemed to have a less porous structure than the NaF (b) group. In the CPP-ACP groups (c) and EMD (d), patch-like structures were clearly seen while some enlarged pores were also observed (Figure 1a-e).

DISCUSSION

This study evaluated the effectiveness of using EMD combined with CPP-ACP on the remineralization of initial caries lesions, and it was found that using the two materials in combination may

have a synergistic effect on biomimetic remineralization. These materials may be proposed as a new treatment alternative for initial caries lesions. However, there is need for further research to clarify this.

The progression of caries or enabling of remineralization depends on the balance between pathological (demineralization) and protective factors (improving remineralization and decreasing bacteria formation) (17). Among the protective factors, fluoride is the most commonly known agent that supports remineralization. Sodium fluoride (NaF), which has a very high water solubility, is the simplest fluoride compound that can be used (18). When toothpaste containing fluoride with an ionic bond (NaF) is used, a CaF_2 layer accumulates on the dental hard tissue while teeth are being brushed. This reserve is used over time, and fluoride concentrations rise in the enamel and saliva (19). A toothpaste containing 500 ppm NaF was used with the positive control group in this study. Controlling the decay process with the fluoride treatment may not be sufficient in individuals who are at a higher risk of caries. Therefore, new anti-caries agents and delivery systems have recently been developed.

EMD is a purified acidic extract of EMP that is secreted from Hertwig's epithelial root sheath in the development of the swine teeth at the embryonic stage (20). EMP that mainly consists of amelogenin, has a hydrophobic structure (21). EMD consist of 90% amelogenin and 10% proline-rich non-amelogenin, tuftelin, tuft protein, serum, ameloblastin, amelin, and saliva proteins (22). EMD regulates enamel biomineralization by inducing and guiding crystal formation, supporting crystal growth, protecting the mineral phase, combining mineral ions, and regulating the growth rate (23).

Wang et al. (24) obtained many new hydroxyapatite crystals using EMP and showed that it plays an important role in hydroxyapatite crystals formation and growth. It was therefore demonstrated that EMP induces enamel remineralization. Chen et al. (25) used EMP to obtain hydroxyapatite nanorods, and thus simulated the enamel mineralization process. These rods were similar to mature enamel not only in terms of their chemical composition, but also in terms of their size. Xiang et al. (26) investigated the remineralization effect of EMDs on initial enamel carious lesions. According to the results of this study, it was found that EMD plays an essential role in promoting the remineralization of initial carious lesions. However, the complete remineralization of such lesions could not be achieved. Considering the findings in these studies, our study aimed to enhance the understanding of the remineralization activity of EMD. Therefore, we used casein phosphopeptide amorphous calcium phosphate, which is a remineralization agent containing Ca and P.

We also used an SEM in this study to determine the structure of the initial carious lesion on the enamel and the changes that occurred after the treatment. Moreover, the changes created by the administered treatment agents at the mineral level in the enamel samples were evaluated using a microanalytical technique with EDX. SEM is a well-defined technique being used in almost every field of modern medicine, and it is generally used as a supportive

technique in various studies. Energy dispersive X-ray spectroscopy (EDS, EDX, or XEDS) is an analytical technique that can be used in combination with SEM for the chemical characterization or elemental analysis of samples (27).

It has been reported in the literature that CPP-ACP remineralization pastes increase the Ca/P% ratio on demineralized enamel surfaces, and thus have the potential of recovering minerals to dental hard tissues (28, 29). Recently, biomineralization of the enamel with a regrown enamel-like mineral layer has been considered in the treatment of initial caries lesions. Self-assembling peptide (30), amelogenin (31), or EMD (26, 32) can be used for this purpose. In our study, SEM-EDX measurements showed that all the groups except the control group exhibited remineralization. It was found that EMD could play an essential role in promoting the remineralization of the initial carious lesions, and previous studies also support this finding (23, 26, 32). However, when used alone, it could not provide complete remineralization of such lesions. CPP-ACP administration provided an effective remineralization of the enamel and therefore, CPP-ACP agent can be an alternative to fluoride in enamel remineralization. The fact that the Ca/P% ratio was higher in the group that received the combination of Emdogain and CPP-ACP compared to the groups that received these materials separately supports the opinion that EMD is an important promoter of enamel bio-remineralization and a modulator of calcium phosphate nanocrystal structures (11). Furthermore, the highest Ca/P% ratio was obtained from the group that received the combination of Emdogain and CPP-ACP. SEM images also supported the data obtained in our study using the EDX microanalytical method.

CONCLUSION

In this study, we found that using a combination of CPP-ACP and EMD could enhance the remineralization activity of each other. However, further studies are required to evaluate the profit-loss relationship and the remineralization activity in terms of the cost, in order to create a standard procedure with regard to the frequency of application, and to evaluate the possible effects on oral-dental and general health in the case of combined CPP-ACP and EMD use.

Ethics Committee Approval: Ethics committee approval was received for this study from the Clinical Research Ethics Committee of İnönü University (2013/146).

Informed Consent: Written informed consent was obtained from participants who participated in this study.

Peer-review: Externally peer-reviewed.

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Conflict of Interest: The authors have no conflicts of interest to declare.

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